Structure-Based Design, Synthesis, and Structure−Activity Relationship Studies of HIV-1 Protease Inhibitors Incorporating Phenyloxazolidinones[†]

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A series of new HIV-1 protease inhibitors with the hydroxyethylamine core and different phenyloxazolidinone P2 ligands were designed and synthesized. Variation of phenyl substitutions at the P2 and P2′ moieties significantly affected the binding affinity and antiviral potency of the inhibitors. In general, compounds with 2- and 4-substituted phenyloxazolidinones at P2 exhibited lower binding affinities than 3-substituted analogues. Crystal structure analyses of ligand−enzyme complexes revealed different binding modes for 2- and 3-substituted P2 moieties in the protease S2 binding pocket, which may explain their different binding affinities. Several compounds with 3-substituted P2 moieties demonstrated picomolar binding affinity and low nanomolar antiviral potency against patient-derived viruses from HIV-1 clades A, B, and C, and most retained potency against drug-resistant viruses. Further optimization of these compounds using structure-based design may lead to the development of novel protease inhibitors with improved activity against drug-resistant strains of HIV-1.

Introduction

The global HIV-AIDS[a] epidemic causes an estimated two million deaths each year and remains one of the most serious healthcare challenges of our time.[1] In the absence of curative or preventative treatments, suppressing HIV-1 replication in infected patients has become a critical goal in managing the disease. Among the anti-HIV drugs developed over the past two decades, inhibitors of HIV-1 reverse transcriptase and protease remain the most prominent in clinical use.[2] These drugs are essential components of highly active antiretroviral therapy (HAART), which typically comprises two reverse transcriptase inhibitors and one protease inhibitor (PI).[3,4] HAART is credited with significantly reducing AIDS-related mortality[5,6] and is currently implemented throughout the world as the standard of care for HIV-AIDS treatment.

The efficacy of HIV-1 PIs has significantly improved since this drug class was first introduced in the mid-1990s.[7] Structure-guided drug design efforts, in both academia and industry, have led to the development of highly potent second-generation PIs with improved potency, tolerability, and pharmacokinetic

profiles. The introduction of low-dose ritonavir as a pharmacokinetic booster in most PI-based regimens has led to the sustained suppression of viral replication for prolonged periods, further improving clinical outcomes.[8−10] Most importantly, the two most recently approved PIs, tipranavir (TPV)[11] and darunavir (DRV),[12−14] retain high antiviral potency against viral strains resistant to most other PIs. Despite these advances, PI therapies are associated with serious problems that significantly limit their effectiveness, including drug side effects, high dosing frequency due to poor pharmacokinetics, and the acquisition of many drug-resistant protease variants.[15]
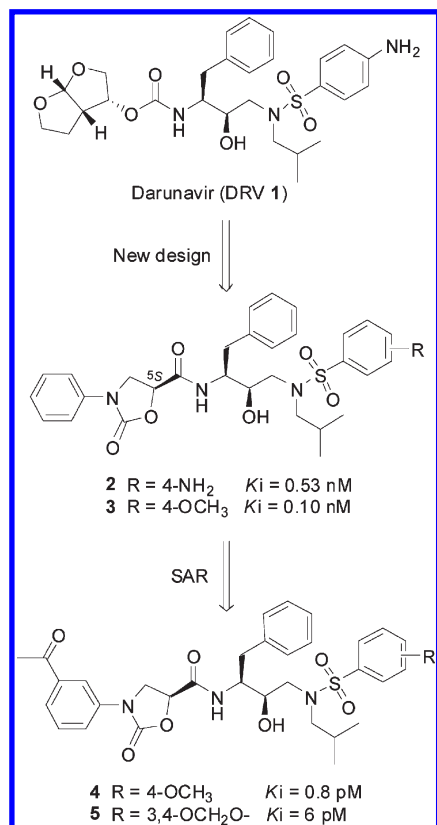
Drug resistance occurs when mutations in the enzyme active site selectively impair drug binding without significantly affecting substrate recognition and processing. The high rate of HIV-1 replication coupled with the infidelity of its reverse transcriptase gives rise to viral quasi-species, which propagate under the selective pressure of drug therapy to confer resistance to these drugs.[16] Alarmingly, even TPV- and DRV-based PI regimens have recently been reported to select multiple resistance mutations in the protease, leading to reduced virological response.[17−19] Therefore, novel PIs must be developed that are less susceptible to drug resistance and maintain broad spectrum activity against multidrug-resistant (MDR) HIV-1 variants with improved pharmacokinetic profiles, providing effective treatments for HIV-AIDS.

To reduce the probability of drug resistance in HIV-1 protease, we have been pursuing a structure-based strategy and recently discovered novel HIV-1 PIs that contain phenyloxazolidinones as P2/P2′ ligands.[20,21] The initial inhibitor design (compounds **2** and **3**, Figure 1) was based on the (*R*)-(hydroxyethylamino)sulfonamide isostere, the core scaffold of APV and DRV, where the P2 tetrahydrofuranyl/bis-tetrahydrofuranyl moiety was replaced with a stereochemically

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Figure 1. Structures of darunavir (DRV) **1** and protease inhibitors **2**−**5**.

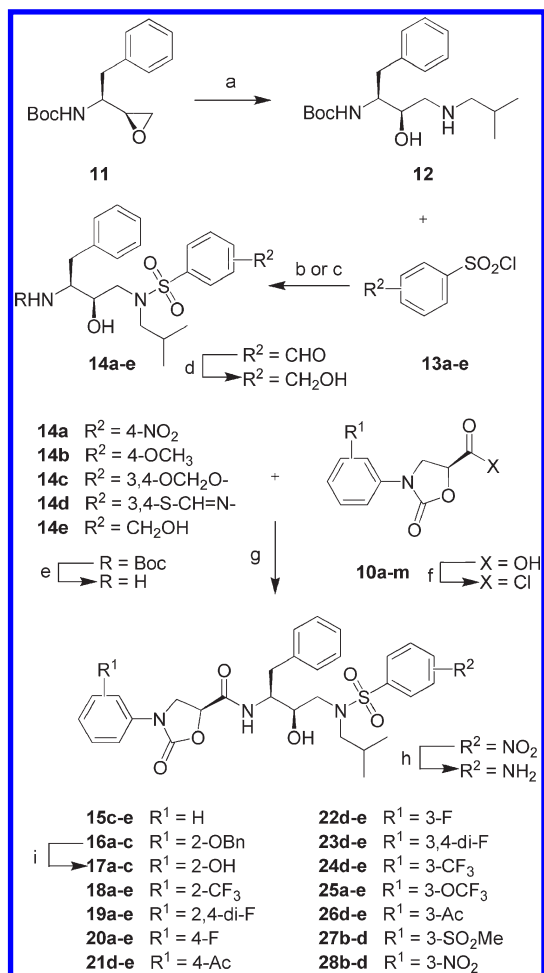**Scheme 1.** Synthesis of (*S*)-*N*-Phenyloxazolidinone-5-carboxylic Acids **10a**−**m**[a]

[a] Reagents and conditions: (a) NaHCO₃, BnOCOCl, (CH₃)₂CO−H₂O (2:1), 0 °C to room temp, overnight; (b) *n*-BuLi, THF, −78 °C to room temp, overnight; (c) NaIO₄, RuCl₃·H₂O, CH₃CN−CCl₄−H₂O (2:2:3), 0 °C to room temp, 4−10 h.

revealing different binding modes for the 2- and 3-substituted P2 moieties in the S2 binding pocket of the enzyme.

Chemistry

The synthesis of enantiomerically pure (*S*)-*N*-phenyloxazolidinone-5-carboxylic acids **10a**−**m** is shown in Scheme 1. Reaction of Cbz-protected aniline derivatives **7a**−**m** with (*S*)-glycidyl butyrate **8** promoted by *n*-BuLi provided enantiomerically pure alcohols (*S*)-5-(hydroxymethyl)-3-phenyloxazolidine-2-ones **9a**−**m**. This one-pot, three-step cascade reaction involves the initial ring-opening of chiral epoxide with *N*-lithium species followed by an intramolecular cyclization and finally an in situ ester hydrolysis.[24] Oxidation of alcohols **9a**−**m** by NaIO₄ and catalytic RuCl₃ provided *N*-phenyloxazolidinone-5-carboxylic acids **10a**−**m** in good overall yields.

The synthesis of protease inhibitor series **15**−**28** incorporating various phenyloxazolidinone-based P2 ligands is illustrated in Scheme 2. The Boc-protected intermediates (*R*)-(hydroxyethylamino)sulfonamides **14a**−**e** were prepared from the commercially available chiral epoxide **11**, (1*S*,2*S*)-(1-oxiranyl-2-phenylethyl)carbamic acid *tert*-butyl ester, using previously developed methods.[20,25] Briefly, ring-opening of epoxide **11** with isobutylamine provided the amino alcohol **12**. Reactions of **12** with sulfonyl chlorides **13a**−**d** in the presence of a base, Na₂CO₃ in a CH₂Cl₂−H₂O mixture for **13a**−**c** and Et₃N in dry CH₂Cl₂ for **13d**, furnished the (*R*)-(hydroxyethylamino)sulfonamide derivatives **14a**−**d**. The 4-(hydroxymethyl)benzenesulfonamide compound **14e** was obtained in two steps, involving reaction of **12** with sulfonyl chloride **13e** followed by reduction of the resulting 4-formylbenzenesulfonamide derivative with NaBH₄ in MeOH. Deprotection of the Boc group in sulfonamides **14a**−**e** followed by reaction of the resulting amines with acid chlorides, obtained by activating the corresponding carboxylic acids **10a**−**m**, afforded the target compound series **15**−**28** (Scheme 2). Inhibitors with the 4-aminobenzenesulfonamide at P2′ were obtained from the corresponding 4-nitrobenzenesulfonamide derivatives by reduction of the nitro group

defined *N*-phenyloxazolidinone-5-carboxamide. Exploration of the P1′ amine, P2 phenyloxazolidinone, and P2′ phenylsulfonamide moieties led to the identification of inhibitors **4** and **5** with exceptionally high binding affinities for wild-type protease (K_i of 0.8 and 6 pM, respectively).[20] Although these binding affinities are comparable to those of lopinavir (LPV) (K_i = 5 pM) and DRV (K_i = 8 pM), compounds **4** and **5** exhibit lower antiviral potencies in cellular assays (EC₅₀ = 30−35 nM)[22] than LPV and DRV.

The structural basis for the high binding affinities of compounds **4** and **5** was revealed by crystal structures of enzyme-inhibitor complexes, involving a network of hydrogen bonds between the inhibitors' oxazolidinone moiety and the invariant Asp29 residue of the protease.[20,23] On the basis of these structures, we sought to enhance the antiviral potency of these compounds by exploring the structure−activity relationships (SAR) at the P2 phenyloxazolidinone and the P2′ phenylsulfonamide moieties. Herein, we report the structure-guided design, synthesis, and biological evaluation of new PIs incorporating a variety of P2 phenyloxazolidinone and P2′ phenylsulfonamide groups into the (*R*)-hydroxyethylamine isostere. The activities of these compounds were tested against wild-type HIV-1 protease and a MDR variant, and their antiviral potencies were determined in cellular assays against patient-derived wild-type viruses and two MDR HIV-1 variants. Several PIs showed low picomolar enzyme inhibitory potencies and high antiviral potencies against wild-type HIV-1 comparable to those of LPV. However, these PIs retained better potencies than LPV against drug-resistant viruses. We also report crystal structures of PIs with 2-substituted phenyloxazolidinones in complex with wild-type protease and compare them with the inhibitor **5**−protease complex structure,

Scheme 2. Synthesis of Protease Inhibitors **15−28**[a]

[a] Reagents and conditions: (a) [i]BuNH₂, EtOH, 80 °C, 3 h; (b) aq Na₂CO₃, CH₂Cl₂, 0 °C to room temp, 6 h; (c) Et₃N, CH₂Cl₂, 0 °C to room temp, 6 h; (d) NaBH₄, MeOH, 0 °C, 30 min; (e) TFA, CH₂Cl₂, room temp, 1 h; (f) (COCl)₂, room temp, overnight; (g) Et₃N, THF, 0 °C to room temp, 6 h; (h) SnCl₂·2H₂O, EtOAc, 70 °C, 3 h; (i) Pd−C, HCO₂NH₄, EtOAc, room temp, overnight.

with SnCl₂·2H₂O. Removal of the benzyl protection in compounds **16a−c** by Pd/C and HCO₂NH₄ in EtOAc provided inhibitors **17a−c**.

The protease inhibitors **29−31** were prepared from the corresponding 3-(nitrophenyl)oxazolidinone analogues **28b−d** as shown in Scheme 3. Reduction of the nitro group in **28b−d** using SnCl₂·2H₂O in EtOAc provided the corresponding inhibitors **29b−d** with 3-(aminophenyl)oxazolidinones as P2 ligands. Further reactions of the 3-aminobenzene compounds **29b−d** with acetic anhydride and methyl chloroformate in CH₂Cl₂ provided the 3-acetamidophenyl (**30b−d**) and methyl 3-phenylcarbamate (**31b−d**) analogues.

Results and Discussion

Analysis of crystal structures of inhibitors **4** and **5** in complex with wild-type HIV-1 protease revealed that the carbonyl oxygen of their oxazolidinone moieties forms direct hydrogen bonds with the amide nitrogen and side chain oxygen atoms of the conserved Asp29 residue in the enzyme S2 binding pocket. In addition, the nitrogen and oxygen atoms of the oxazolidinone ring form water-mediated hydrogen bonds with the carbonyl oxygen of Gly48 in the protease flap and the backbone amide

Scheme 3. Synthesis of Protease Inhibitors **28−31**[a]
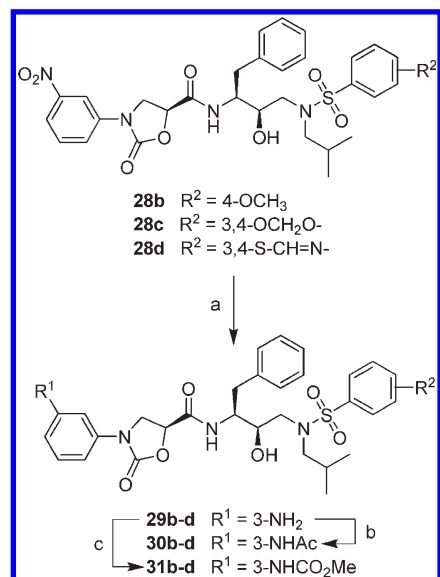
[a] Reagents and conditions: (a) SnCl₂·2H₂O, EtOAc, 70 °C, 3 h; (b) Ac₂O, CH₂Cl₂, room temp, 2 h; (c) MeOCOCl, Na₂CO₃, CH₂Cl₂, 0 °C to room temp, 2 h.

nitrogen of Asp30, respectively.[20,23] However, unlike DRV,[14] inhibitors **4** and **5** do not directly interact with the backbone or side chain atoms of Asp30. Since the 2-position of the phenyl ring is observed close to Asp30, we reasoned that a substitution at this position might directly interact with the Asp30 residue of protease. On the basis of these observations, we designed a series of new HIV-1 protease inhibitors by incorporating a variety of 2-, 3-, and 4-substituted (S)-N-phenyloxazolidinones into the (R)-(hydroxyethylamino)sulfonamide isostere. Variations at the P2 phenyloxazolidinone moiety were explored in combination with five different phenylsulfonamide groups as P2′ ligands.

Structure−Activity Relationships. The enzyme inhibitory potencies of all compounds were measured against wild-type HIV-1 protease (Q7K) and a drug-resistant variant (L10I, L63P, A71V, G73S, I84V, L90M) using fluorometric assays as previously described (Table 1).[20,25] Selected compounds were further tested by Monogram Biosciences for antiviral activity in cellular assays against three patient-derived wild-type viruses from predominant HIV-1 clades (A, B, and C) and against a drug-resistant virus with protease mutations M46I, I54V, V82A, and L90M. LPV and DRV were used as drug controls, and a HIV-1 NL4-3-derived clade B laboratory strain (CNDO) and a MDR strain (MDRC4) were used as virus controls (Table 2).

The first series of inhibitors **15c−e**, incorporating the unsubstituted phenyloxazolidinone moiety at the P2 position and various benzenesulfonamides at the P2′ position, displayed subnanomolar inhibitory potency against wild-type protease. The 1,3-benzodioxolane and 4-(hydroxymethyl)-benzene analogues **15c** and **15e** are slightly less potent than inhibitors **2** and **3**, but the benzothiazole compound **15d** is more potent against both wild-type (K_i = 0.033 nM) and MDR (K_i = 1.21 nM) proteases. In cellular assays, **15d** inhibited HIV-1 viral replication with EC₅₀ values of 14−41 nM against wild-type viruses but lost about 6-fold potency against MDR viruses.

To test if 2-position substitutions at the P2 phenyloxazolidinone moiety could improve binding affinity to protease,

Table 1. Inhibitory Activity of Compounds against Wild-Type HIV-1 Protease and an MDR Variant[a]
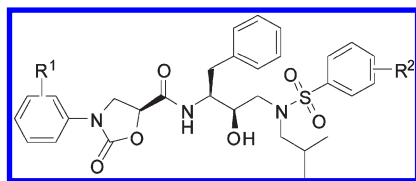
compd	R^1	R^2	K_i (nM) Wt	MDR
15c	H	3,4-OCH2O−	0.206	5.50
15d	H	3,4-S−C=N−	0.033	1.21
15e	H	4-CH2OH	0.252	NT
17a	2-OH	4-NH2	7.36	40.46
17b	2-OH	4-OCH3	5.65	67.34
17c	2-OH	3,4-OCH2O−	1.47	NT
18a	2-CF3	4-NH2	0.136	5.71
18b	2-CF3	4-OCH3	0.086	2.45
18c	2-CF3	3,4-OCH2O−	0.142	4.36
18d	2-CF3	3,4-S−C=N−	0.097	1.86
18e	2-CF3	4-CH2OH	0.235	4.57
19a	2,4-di-F	4-NH2	0.385	NT
19b	2,4-di-F	4-OCH3	0.063	10.54
19c	2,4-di-F	3,4-OCH2O−	0.347	NT
19d	2,4-di-F	3,4-S−C=N−	0.150	3.50
19e	2,4-di-F	4-CH2OH	0.212	NT
20a	4-F	4-NH2	0.448	NT
20b	4-F	4-OCH3	0.128	9.57
20c	4-F	3,4-OCH2O−	0.167	10.78
20d	4-F	3,4-S−C=N−	0.133	3.0
20e	4-F	4-CH2OH	0.207	9.56
21d	4-Ac	3,4-S−C=N−	0.073	2.04
21e	4-Ac	4-CH2OH	0.317	NT
22d	3-F	3,4-S−C=N−	0.080	2.12
22e	3-F	4-CH2OH	0.319	NT
23d	3,4-di-F	3,4-S−C=N−	0.232	5.18
23e	3,4-di-F	4-CH2OH	0.330	NT
24d	3-CF3	3,4-S−C=N−	0.016	2.96
24e	3-CF3	4-CH2OH	0.196	10.15
25a	3-OCF3	4-NH2	0.225	9.95
25b	3-OCF3	4-OCH3	0.130	16.3
25c	3-OCF3	3,4-OCH2O−	0.239	10.6
25d	3-OCF3	3,4-S−C=N−	0.026	3.38
25e	3-OCF3	4-CH2OH	0.286	7.49
26d	3-Ac	3,4-S−C=N−	0.015	1.69
26e	3-Ac	4-CH2OH	0.236	NT
27b	3-SO2CH3	4-OCH3	0.049	1.74
27c	3-SO2CH3	3,4-OCH2O−	0.025	4.16
27d	3-SO2CH3	3,4-S−C=N−	0.003	2.45
28b	3-NO2	4-OCH3	0.136	5.40
28c	3-NO2	3,4-OCH2O−	0.117	6.48
28d	3-NO2	3,4-S−C=N−	0.015	0.93
29b	3-NH2	4-OCH3	0.020	2.68
29c	3-NH2	3,4-OCH2O−	0.113	6.55
29d	3-NH2	3,4-S−C=N−	0.008	1.84
30b	3-NHAc	4-OCH3	0.122	3.89
30c	3-NHAc	3,4-OCH2O−	0.051	4.76
30d	3-NHAc	3,4-S−C=N−	0.006	2.83
31b	3-NHCO2CH3	4-OCH3	0.163	4.44
31c	3-NHCO2CH3	3,4-OCH2O−	0.289	3.68
31d	3-NHCO2CH3	3,4-S−C=N−	0.026	1.87
LPV			0.005	0.90
DRV			0.008	0.025

[a]Wt: Q7K. MDR: L10I, L63P, A71V, G73S, I84V, L90M. NT = not tested.

we prepared inhibitors **17−19** with 2-hydroxy, 2-trifluoromethyl, and 2,4-difluoro substitutions on the phenyl ring

that were designed to interact with the Asp30 residue of the protease. Surprisingly, protease inhibitors containing the 2-hydroxyphenyl group at the P2 moiety **17a−c** exhibited significantly lower inhibitory potency than the corresponding unsubstituted analogues (**2**, **3**, and **15c**), with K_i values of all three analogues in the nanomolar range. However, inhibitors with the 2-trifluoromethylphenyl (**18a−e**) and 2,4-difluorophenyl (**19a−e**) groups were equipotent to inhibitors with unsubstituted phenyl rings. Similarly, compounds with the 4-fluorophenyl (**20a−e**) and 4-acetylphenyl (**21d−e**) groups on the P2 moiety did not have better potency than the corresponding unsubstituted phenyl compounds. In each series, inhibitors with the 4-methoxybenzene and benzothiazole moieties at P2′ position are more potent than compounds with other groups.

Since compounds with the 2-substituted phenyl group at the P2 moiety were generally less potent than the corresponding unsubstituted analogues, we explored various 3-substituted phenyl groups at the P2 oxazolidinone moiety. Compounds **22** and **23** with the 3-fluorophenyl and 3,4-difluorophenyl groups at the P2 moiety showed inhibitory potency in the subnanomolar range, with analogues **22d** and **23d** containing benzothiazole moieties at P2′ being the most potent. Introducing a 3-trifluoromethylphenyl group at the P2 markedly improved potency, as previously observed;[20] compound **24d** with the benzothiazole group at P2′ exhibited a K_i of 16 pM. Replacing the 3-trifluoromethylphenyl group with a 3-trifluoromethoxyphenyl (compounds **25a−e**) failed to improve potency; only the P2′ benzothiazole inhibitor **25d** showed inhibitory potency in the picomolar range (K_i = 0.026 nM).

Incorporating the 3-acetylphenyloxazolidinone as the P2 ligand in combination with the 4-methoxyphenyl and 1,3-benzodioxolane groups at the P2′ position has previously provided highly potent protease inhibitors such as **4** and **5**.[20] However, these compounds showed lower antiviral potency in cellular assays than LPV and DRV, presumably because of their higher lipophilicity (cLogP > 5 for both compounds). Combining this high affinity P2 ligand with the more polar benzothiazole group at P2′ position provided the analogue **26d** with similarly high inhibitory potency against wild-type (K_i = 15 pM) and MDR (K_i = 1.69 nM) proteases. Compared to inhibitors **4** and **5**, compound **26d** exhibited better antiviral potency against wild-type viruses (EC_{50} = 4.2−10.1 nM), though it lost 5-fold potency against MDR viruses.

On the basis of these encouraging results, we explored various 3-substituted phenyloxazolidinones that could mimic the key interactions of the 3-acetylphenyloxazolidinone moiety of **4** and **5**. Replacing the 3-acetyl group on the phenyl ring with an isosteric 3-methylsulfonyl moiety greatly improved the cellular antiviral potency of resulting analogues **27b−d**. Thus, compounds with the P2′ 4-methoxyphenyl (**27b**, K_i = 49 pM) and 1,3-benzodioxolane (**27c**, K_i = 25 pM) groups showed high antiviral potency against wild-type and MDR viruses (EC_{50} of 4.7−18.6 and 3.5−13.6 nM, respectively). Analogue **27d** with the P2′ benzothiazole group showed the best binding affinity (K_i = 3 pM) and high cellular antiviral potency against wild-type viruses (EC_{50} = 3.5−5.7 nM) and retained significant potency against MDR viruses (EC_{50} = 9.4−10.5 nM). Interestingly, **27c** and **27d** retain antiviral potencies against MDR viruses and have relatively flat potency profiles against both wild-type and MDR viruses, while LPV, which is more potent against wild-type viruses (EC_{50} = 1.4−1.6 nM), displayed

Table 2. Antiviral Activity of Compounds against Wild-Type and MDR HIV-1 Strains[a]
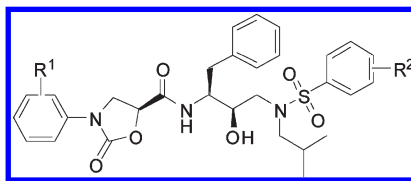
compd	R^1	R^2	EC$_{50}$ (nM) (fold change)					
			WT	WT-A	WT-B	WT-C	MDR1	MDR
15d	H	3,4-S−C=N−	21.5	29.8	14.1	40.9	112.9 (5)	123.1 (6)
24d	3-CF$_3$	3,4-S−C=N−	13.5	16.1	9.9	32.5	57.6 (4)	53.4 (4)
25d	3-OCF$_3$	3,4-S−C=N−	22.1	32.5	12.7	57.8	104.5 (5)	110.8 (5)
26d	3-Ac	3,4-S−C=N−	6.5	9.7	4.2	10.1	32.3 (5)	34.7 (5)
27b	3-SO$_2$CH$_3$	4-OCH$_3$	5.5	7.3	4.7	9.6	14.4 (3)	18.6 (3)
27c	3-SO$_2$CH$_3$	3,4-OCH$_2$O−	4.0	7.8	3.5	7.6	13.6 (3)	12.1 (3)
27d	3-SO$_2$CH$_3$	3,4-S−C=N−	5.0	4.1	3.5	5.7	9.4 (2)	10.5 (2)
28d	3-NO$_2$	3,4-S−C=N−	7.3	11.5	5.7	16.3	33.8 (3)	37.7 (5)
29b	3-NH$_2$	4-OCH$_3$	4.0	8.5	3.0	9.4	19.0 (5)	36.4 (9)
29c	3-NH$_2$	3,4-OCH$_2$O-	7.1	6.1	2.8	12.3	27.9 (4)	35.4 (5)
29d	3-NH$_2$	3,4-S−C=N−	1.7	2.6	1.3	2.5	11.9 (7)	14.6 (9)
30c	3-NHAc	3,4-OCH$_2$O-	10.9	9.0	6.3	21.1	41.9 (4)	46.9 (4)
30d	3-NHAc	3,4-S−C=N−	5.8	10.1	7.7	9.5	25.4 (4)	29.5 (5)
31d	3-NHCO$_2$CH$_3$	3,4-S−C=N−	1.2	2.4	1.6	3.0	10.8 (9)	13.0 (11)
LPV			1.4	1.6	1.0	1.6	34.1 (24)	39.7 (28)
DRV			0.4	0.6	0.4	0.4	0.5 (1)	1.0 (2)

[a] Antiviral assays were carried out by Monogram Biosciences. WT, wild-type HIV-1 control; WT-A, WT-B, WT-C, patient-derived strains of wild-type HIV-1 from clades A, B, and C, respectively; MDR1, drug-resistant HIV-1 variant with protease mutations M46I, I54V, V82A, and L90M; MDR, drug-resistant HIV-1 control MDRC4. Values in the parentheses represent the change in EC$_{50}$ values compared to the WT control.

significantly lower potency against MDR viruses (EC$_{50}$ = 34.1−39.7 nM).

To further manipulate the 3-position of the phenyl ring using simple transformations, we next introduced a nitro group at this position. Among the compounds with the 3-nitrophenyl group, **28b−d**, the benzothiazole analogue **28d** exhibited potent inhibitory activity (K_i = 15 pM) and good antiviral potency in cellular assays (EC$_{50}$ = 5.7−37.7 nM) against the viruses tested. Reduction of the nitro group provided the 3-aminophenyl analogues **29b−d** with better inhibitory potencies than those of the parent nitrophenyl compounds, and all three potently inhibited HIV-1 replication. The analogues **29b** and **29d** (K_i of 20 and 8 pM, respectively), with the 4-methoxyphenyl and benzothiazole groups at P2′, showed low nanomolar antiviral potency; **29d** (EC$_{50}$ = 1.7 nM) was equipotent to LPV against wild-type viruses but retained better potency against the MDR variants (EC$_{50}$ = 11.9−14.6 nM).

Further transformation of the 3-aminophenyl analogues **29b−d** to the 3-acetamidophenyl and methyl 3-phenylcarbamate derivatives did not significantly improve the inhibitory or antiviral potency of resulting compounds **30b−d** and **31b−d**. However, the P2′ benzothiazole analogues **30d** and **31d** showed highly potent activity (K_i of 6 and 26 pM, respectively) in both enzymatic and cellular assays. The antiviral potency of the carbamate analogue **31d** (EC$_{50}$ = 1.2−3.0 nM) is comparable to that of LPV against wild-type viruses but 3-fold better against the MDR HIV-1 variants.

These findings indicate that the antiviral activity in this series can be significantly improved by changing substitutions at the P2 phenyloxazolidinone and P2′ phenylsulfonamide moieties, and these changes can also improve the resistance profile of inhibitors. Analogues **27d**, **29d**, and **31d** all have the same benzothiazole moiety at P2′, but variations at the phenyl ring of the P2 moiety resulted in different activity profiles against wild-type and MDR viruses. Compound **27d**
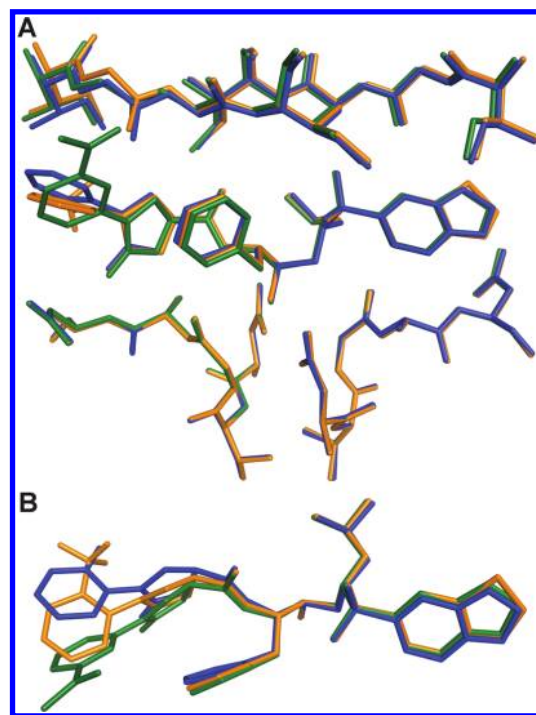
is slightly less potent than LPV against wild-type viruses, but it lost only 2-fold potency against the two MDR viruses tested, whereas LPV lost 24- to 28-fold potency against the same viruses. On the other hand, compounds **29d** and **31d** inhibited wild-type viruses with high potency comparable to that of LPV but lost relatively more potency against MDR viruses, though they retained better potency than LPV. Although the new PIs contain the same (*R*)-(hydroxyethylamino)sulfonamide core as DRV and exhibit similar binding affinities to protease, their antiviral potencies against both wild-type and MDR viruses are lower than that of DRV. Further optimization of these PIs may lead to compounds with improved antiviral potencies and better resistance profiles.

Crystal Structures of Protease Complexes. To understand why the observed protease inhibitory activities were significantly different for compounds with 2- and 3-substituted P2 phenyloxazolidinone moieties, we determined crystal structures of **17c** and **18d** in complex with the wild-type HIV-1 protease and compared them with the structure of the inhibitor **5**−protease complex (Figure 2).[23] The three complexes differ mainly at the P2 phenyloxazolidinone moieties where they have different substituent groups at the 2- (**17c** and **18d**) and 3-position (**5**) on the phenyl ring; **18d** also has a different P2′ moiety (benzothiazole). These differences at the P2 moiety resulted in the phenyl rings adopting different orientations in the three ligand−enzyme complexes; the phenyl ring in **17c** and **18d** is perpendicular to the oxazolidinone ring, whereas both rings in **5** are almost in the same plane (Figure 3). As a result, all three structures show variations in the backbone near one of the flap regions where the inhibitor's phenyloxazolidinone moiety interacts with the protease. Other than these differences, the three structures superpose well onto each other with a maximum rmsd of 0.264 for all Cα atoms.
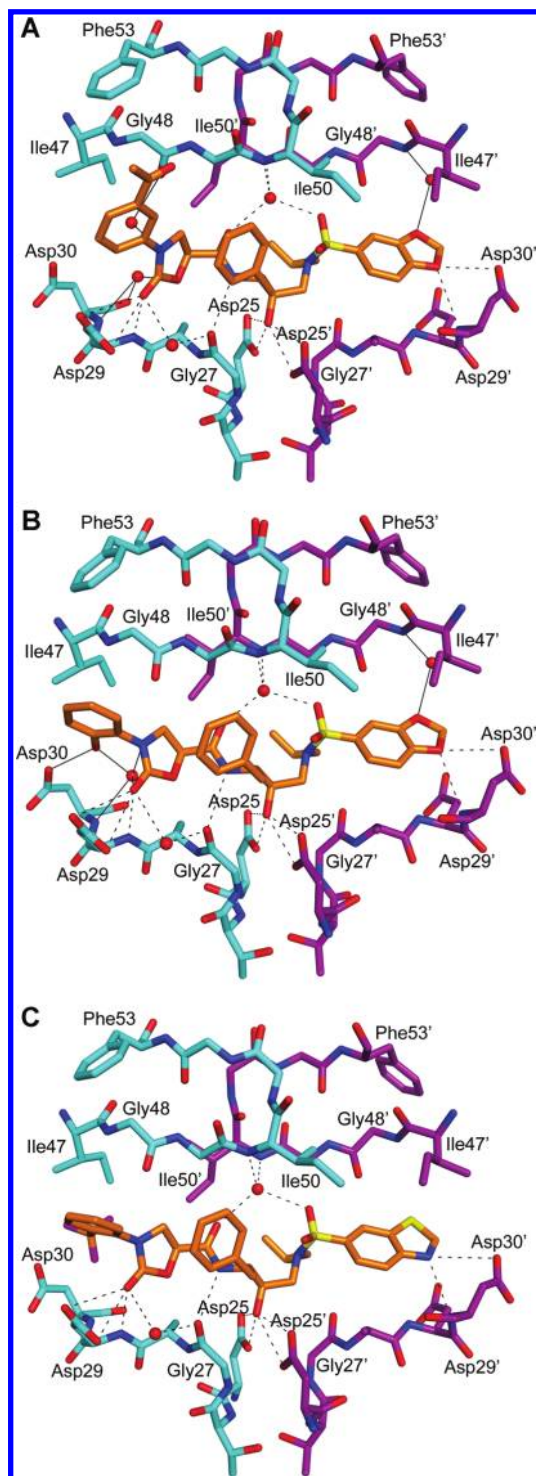
Figure 2. Crystal structures of inhibitors **5** (A), **17c** (B), and **18d** (C) in complex with HIV-1 protease. Dashed lines represent hydrogen bonds common to all three inhibitors, and solid lines represent hydrogen bonds specific to each inhibitor.

Figure 3. Superposition of protease−inhibitor complexes of **5** (green), **17c** (blue), and **18d** (orange): inhibitors in the protease active site (A) and inhibitors only (B).

The conformational changes at the P2 phenyloxazolidinone moiety resulted in different hydrogen bonding patterns and van der Waals (vdW) interactions in all three complexes. The nitrogen atom of the oxazolidinone moiety forms a water-mediated hydrogen bond with the oxygen atom of the Gly48 in **5** and a water-mediated hydrogen bond with the nitrogen atom of Asp30 in **17c**, whereas no water-mediated hydrogen bond is associated with this nitrogen atom in the

crystal structure of **18d**. The acetyl group on the phenyl ring in **5** does not form any hydrogen bond with the protease but forms vdW contacts with the residues Gly48 and Gly49 as well as the side chain of Phe53 in the flap region. The hydroxyl group on the phenyl ring in **17c** forms a direct hydrogen bond with the side chain of Asp30 and a water-mediated hydrogen bond with the main chain nitrogen of Asp30. Although **17c** forms more hydrogen bonds than **5**, the binding affinity of **17c** is significantly lower than that of **5**, emphasizing that the binding affinity of an inhibitor is a product of both enthalpic interactions, including hydrophilic and hydrophobic contributions, and the entropy of binding.[26,27]

HIV-1 protease is known to be highly plastic and can adopt different conformations when binding to various ligands.[23] In these three complexes, however, the conformational changes in the ligands are more pronounced than those in the protease. These conformational differences at P2 moieties alter the compounds' interactions with the protease in the S2 binding pocket, which likely account for their significantly different binding affinities.

Conclusion

We have evaluated the structure−activity profiles of a series of HIV-1 protease inhibitors containing phenyloxazolidinone-based P2 ligands to improve the cellular antiviral potency. On the basis of the structures of ligand−enzyme complexes, a number of (*S*)-*N*-phenyloxazolidinone-5-carboxylic acids were designed, synthesized, and incorporated into the (*R*)-hydroxyethylamine isostere as P2 ligands in combination with several phenylsulfonamides as P2′ ligands. In nearly all cases, compounds with the 2-substituted phenyloxazolidinone moieties, designed to improve binding in the S2 region of the enzyme, exhibited lower binding affinities than the 3-substituted analogues. Comparison of the crystal structures of PIs with 2- and 3-substituted P2 phenyloxazolidinone moieties

revealed different orientations of the P2 moieties in the S2 binding pocket of the protease, possibly explaining the reduced potencies of the 2-substituted analogues.

Our SAR studies show that P2 moieties with polar substituents at the 3-position of the phenyl ring significantly enhance antiviral potency in the resulting compounds. The polarity and size of the P2′ moiety also strongly influenced the enzymatic and antiviral potencies; compounds with large polar groups at P2′ have better potencies than those with smaller groups. In each series, compounds with the benzothiazole moiety at P2′ were the most potent, exhibiting the highest enzymatic and antiviral potencies. These efforts identified three compounds, **27d**, **29d**, and **31d**, with the benzothiazole moiety at P2′ and different phenyloxazolidinone moieties at P2, which demonstrated excellent antiviral potencies against wild-type viruses and better potencies against MDR viruses than LPV. Since variations at the P2 and P2′ moieties significantly affected the antiviral activities of compounds against wild-type and MDR viruses, further optimization of these groups may lead to PIs with improved antiviral potencies against MDR HIV-1 variants.

Experimental Section

General Chemistry. Proton (^{1}H NMR) and carbon nuclear magnetic resonance (^{13}C NMR) spectra were recorded with a Varian Mercury 400 MHz NMR spectrometer operating at 400 MHz for ^{1}H and 100 MHz for ^{13}C NMR. Chemical shifts are reported in ppm (δ scale) relative to the solvent signal, and coupling constants (J) are reported in hertz. Data are represented as follows: chemical shift, multiplicity (s) singlet, (d) doublet, (t) triplet, (q) quartet, (m) multiplet, (br) broad, (dd) doublet of doublet), coupling constant in Hz, and integration. High-resolution mass spectra (HRMS) were recorded on a Waters Q-TOF Premier mass spectrometer by direct infusion of solutions of each compound using electrospray ionization (ESI) in the positive mode. cLogP values were calculated using ChemBioDraw (CambridgeSoft), version 11. All reactions were performed in oven-dried round-bottomed or modified Schlenk flasks fitted with rubber septa under dry N_2 atmosphere, unless otherwise noted. Flash and column chromatography was performed using silica gel (230−400 mesh, Merck KGA). Analytical thin-layer chromatography (TLC) was performed using silica gel (60 F-254) coated aluminum plates (Merck KGA), and spots were visualized by exposure to ultraviolet light (UV) and/or exposure to an acidic solution of *p*-anisaldehyde (anisaldehyde) followed by brief heating. Dichloromethane was dried over P_2O_5 and distilled; tetrahydrofuran (THF) was distilled from sodium/ benzophenone. All other chemicals, reagents, and solvents were purchased from commercial sources and used as received.

The purity of all final compounds was determined by analytical reversed-phase high performance liquid chromatography (HPLC) using two different chromatographic systems and was found to be ≥95%. Analytical HPLC was performed on a Waters Separation module 2695 system equipped with an autosampler and a Waters 996 photodiode array detector. The first system consisted of the following: column, Waters XTerra RP-C18 (3.5 μm, 4.6 mm × 150 mm); mobile phase A, 10 mM ammonium acetate in water; mobile phase B, acetonitrile. At a flow rate of 1.0 mL/min, gradient elution was performed from 50% B to 100% B over 10 min. The second system consisted of the following: column, Alltech Prevail RP-C18 (3.0 μm, 4.6 mm × 150 mm); mobile phase A, 10 mM ammonium acetate in water; mobile phase B, acetonitrile. Gradient elution was performed from 50% B to 100% B over 10 min at a flow rate of 1.0 mL/min. HPLC retention times and purity data for each target compound are provided in the Supporting Information.

Typical Procedures for the Synthesis of Protease Inhibitors. (*S*)-5-(Hydroxymethyl)-3-(3-nitrophenyl)oxazolidin-2-one (9m). A solution of the Cbz-protected aniline derivative **6m** (9.45 g, 34.7 mmol) in dry THF (150 mL) was cooled to −78 °C under dry nitrogen atmosphere. A solution of *n*-BuLi (1.6 M in hexanes; 25 mL, 40 mmol) was slowly added, keeping the reaction temperature below −70 °C. After the reaction mixture was stirred at −78 °C for 45 min, a solution of (*S*)-glycidyl butyrate **7** (5 g, 34.7 mmol) in dry THF was slowly added. The resulting mixture was stirred at −78 °C for 2 h and then slowly warmed to room temperature and stirred overnight. Reaction was quenched by the addition of saturated aqueous NH$_4$Cl solution (50 mL). EtOAc (200 mL) and water (50 mL) were added and layers separated. The aqueous layer was further extracted with EtOAc (3 × 150 mL). Combined organic extract was washed with saturated aqueous NaCl solution (100 mL), dried (Na$_2$SO$_4$), filtered, and evaporated under reduced pressure to yield a pale yellow solid. This solid was triturated with a mixture of chloroform and hexanes (1:3) (50 mL) and filtered to provide pure alcohol **9m** (5.2 g, 63%) as a pale yellow solid. ^{1}H NMR (400 MHz, CDCl$_3$) δ 8.27 (t, J = 2.0 Hz, 1H), 8.13 (m, 1H), 7.99 (m, 1H), 7.55 (t, J = 8.4 Hz, 1H), 4.82 (m, 1H), 4.15−4.04 (m, 3H), 3.80 (dd, J = 8.4, 6.0 Hz, 1H), 2.18 (br s, 1H); MS (ES+) *m/z* 239.1 [M + H]$^+$.

(*S*)-3-(3-Nitrophenyl)-2-oxooxazolidine-5-carboxylic Acid (10m). To an ice-cooled solution of NaIO$_4$ (35 mmol) in water (75 mL) was added a solution of the alcohol **9m** (2.4 g, 10 mmol) in a mixture of CH$_3$CN and CCl$_4$ (1:1) (100 mL). Solid RuCl$_3$·H$_2$O (0.105 g, 0.5 mmol) was added, and the reaction mixture was stirred at 0 °C for 30 min, warmed to room temperature, and stirred for 6 h. Reaction was quenched by adding CH$_2$Cl$_2$ (100 mL). Layers were separated, and the aqueous layer was further extracted with CH$_2$Cl$_2$ (2 × 100 mL). The combined organic extract was dried (Na$_2$SO$_4$), filtered, and evaporated. The residue was purified by column chromatography on silica gel, eluting with 30% CH$_3$CN in CH$_2$Cl$_2$ + 1% HCO$_2$H to provide the acid **10m** (2.1 g, 83%) as a pale yellow solid. ^{1}H NMR (400 MHz, CD$_3$OD) δ 8.59 (t, J = 2.4 Hz, 1H), 8.01 (ddd, J = 8.0, 2.4, 1.2 Hz, 1H), 7.91 (ddd, J = 8.0, 2.4, 1.2 Hz, 1H), 7.63 (t, J = 8.0 Hz, 1H), 5.21 (dd, J = 9.6, 5.2 Hz, 1H), 4.47 (t, J = 9.2 Hz, 1H), 4.24 (dd, J = 9.6, 5.6 Hz, 1H); MS (ES−) *m/z* 251.1 [M − H]$^-$.

(5*S*)-*N*-[(1*S*,2*R*)-3-[(6-Benzothiazolylsulfonyl)(2-methylpropyl)-amino]-2-hydroxy-1-(phenylmethyl)propyl]-3-(3-nitrophenyl)-2-oxooxazolidine-5-carboxamide (28d). Excess oxalyl chloride (6 mL) was added to the acid **10m** (0.253 g, 1.0 mmol), and the resulting mixture was stirred at room temperature overnight. Oxalyl chloride was removed by distillation under reduced pressure, and the residue was dried under high vacuum for 30 min. The resulting acid chloride was dissolved in dry THF (10 mL) and was used in the coupling reaction.

To a solution of the *N*-[(1*S*,2*R*)-3-[(6-benzothiazolylsulfonyl)- (2-methylpropyl)amino]-2-hydroxy-1-(phenylmethyl)propyl]- carbamic acid-1,1-dimethylethyl ester[25] **14d** (0.534 g, 1 mmol) in CH$_2$Cl$_2$ (10 mL) was added trifluoroacetic acid (4 mL), and the resulting mixture was stirred at room temperature for 1 h. The reaction mixture was concentrated under reduced pressure. The residue was dissolved in toluene (10 mL) and evaporated at reduced pressure to dryness. The resulting deprotected amine was dissolved in dry THF (10 mL), and the solution was cooled 0 °C. Et$_3$N (0.3 mL, 2.15 mmol) was added, and after 15 min the above acid chloride solution was slowly added. The resulting reaction mixture was stirred at 0 °C for 15 min, allowed to warm to room temperature, and stirred until reaction was complete (monitored by TLC). Water (10 mL) and EtOAc (50 mL) were added, and layers were separated. The organic portion was washed with saturated aqueous NaCl solution (30 mL), dried (Na$_2$SO$_4$), filtered, and evaporated. The residue was purified by flash chromatography on silica gel, eluting with a mixture of EtOAc−hexanes (4:2) to afford the target compound **28d** (0.56 g, 84%) as a white solid. ^{1}H NMR (400 MHz, CDCl$_3$) δ 9.23 (s, 1H), 8.50 (d, J = 1.6 Hz, 1H), 8.28 (d, J = 8.4 Hz, 1H), 8.22

(t, J = 2.0 Hz, 1H), 8.05 (m, 1H), 7.93−7.90 (m, 2H), 7.60 (t, J = 8.4 Hz, 1H), 7.13 (d, J = 7.2 Hz, 2H), 7.02 (t, J = 7.2 Hz, 2H), 6.85 (t, J = 7.2 Hz, 1H), 6.79 (d, J = 10.0 Hz, 1H), 4.84 (dd, J = 9.6, 5.2 Hz, 1H), 4.29 (m, 1H), 4.09 (t, J = 9.2 Hz, 1H), 3.99 (m, 1H), 3.41 (dd, J = 9.2, 6.0 Hz, 1H), 3.29 (dd, J = 15.6, 9.6 Hz, 1H), 3.14−3.05 (m, 3H), 2.93 (dd, J = 13.2, 6.4 Hz, 1H), 2.78 (dd, J = 13.6, 10.4 Hz, 1H), 1.89 (m, 1H), 0.96 (d, J = 6.8 Hz, 3H), 0.91 (d, J = 6.4 Hz, 3H); ^{13}C NMR (100 MHz, CDCl$_3$) δ 168.36, 158.38, 155.90, 152.82, 148.86, 138.69, 137.44, 135.68, 134.69, 130.33, 129.61 (2C), 128.62 (2C), 126.67, 125.05, 124.75, 123.92, 122.59, 119.35, 113.01, 72.58, 69.98, 59.12, 53.91, 53.45, 48.10, 35.53, 27.57, 20.38, 20.13. HRMS (ESI) m/z calcd for C$_{31}$H$_{34}$N$_5$O$_8$S [M + H]$^+$ 668.1849; found 668.1852.

(5S)-3-(3-Aminophenyl)-N-[(1S,2R)-3-[(6-benzothiazolylsulfonyl)-(2-methylpropyl)amino]-2-hydroxy-1-(phenylmethyl)propyl]-2-oxooxazolidine-5-carboxamide (29d). A mixture of the above nitro compound **28d** (0.335 g, 0.5 mmol) and SnCl$_2$.2H$_2$O (0.565 g, 2.5 mmol) in EtOAc (20 mL) was heated at 80 °C for 3 h. The reaction mixture was allowed to cool to ambient temperature and treated with saturated aqueous NaHCO$_3$ solution (15 mL). The mixture was diluted with EtOAc (50 mL), and layers were separated. The aqueous layer was further extracted with EtOAc (2 × 50 mL). The combined organic extract was washed with saturated aqueous NaCl solution (2 × 40 mL), dried (Na$_2$SO$_4$), filtered, and evaporated. The residue was purified by flash chromatography on silica gel, eluting with a mixture of 1% MeOH in EtOAc to provide the target compound **29d** (0.29 g, 91%) as a white solid. ^{1}H NMR (400 MHz, CDCl$_3$) δ 9.21 (s, 1H), 8.49 (d, J = 2.0 Hz, 1H), 8.27 (d, J = 8.8 Hz, 1H), 7.92 (dd, J = 8.8, 1.6 Hz, 1H), 7.17−7.12 (m, 3H), 7.09−7.04 (m, 3H), 6.96 (t, J = 7.6 Hz, 1H), 6.90 (d, J = 9.6 Hz, 1H), 6.55 (dd, J = 8.0, 1.6 Hz, 1H), 6.51 (dd, J = 8.0, 2.0 Hz, 1H), 4.73 (dd, J = 10.0, 6.4 Hz, 1H), 4.21 (m, 1H), 4.0 (t, J = 9.6 Hz, 2H), 3.85 (br s, 1H), 3.75 (br s, 1H), 3.32 (dd, J = 9.6, 6.4 Hz, 1H), 3.25 (dd, J = 15.2, 9.2 Hz, 1H), 3.14 (dd, J = 14.0, 4.8 Hz, 1H), 3.09−3.03 (m, 2H), 2.93 (dd, J = 13.6, 6.8 Hz, 1H), 2.73 (dd, J = 13.6, 10.8 Hz, 1H), 1.89 (m, 1H), 0.95 (d, J = 6.4 Hz, 3H), 0.91 (d, J = 6.4 Hz, 3H); ^{13}C NMR (100 MHz, CDCl$_3$) δ 169.07, 158.34, 155.87, 153.14, 147.55, 138.46, 137.47, 135.69, 134.66, 130.10, 129.53 (2C), 128.70 (2C), 126.85, 125.09, 124.72, 122.60, 111.81, 108.37, 105.48, 72.49, 68.87, 59.11, 53.99, 53.74, 48.47, 35.84, 27.55, 20.38, 20.12. HRMS (ESI) m/z calcd for C$_{31}$H$_{36}$N$_5$O$_6$S$_2$ [M + H]$^+$ 638.2107; found 638.2098.

(5S)-3-[3-(Acetylamino)phenyl]-N-[(1S,2R)-3-[(6-benzothiazolylsulfonyl)(2-methylpropyl)amino]-2-hydroxy-1-(phenylmethyl)propyl]-2-oxooxazolidine-5-carboxamide (30d). To a solution of compound **29d** (0.1 g, 0.157 mmol) in dry CH$_2$Cl$_2$ was added Ac$_2$O (30 μL), and the resulting solution was stirred at room temperature for 2 h. The solvents were evaporated under reduced pressure and the residue was purified by flash chromatography, eluting with 1% MeOH in EtOAc to afford the title compound **30d** (0.1 g, 94%) as a white solid. ^{1}H NMR (400 MHz, CDCl$_3$) δ 9.20 (s, 1H), 8.49 (d, J = 2.0 Hz, 1H), 8.23 (d, J = 8.8 Hz, 1H), 7.95 (s, 1H), 7.91 (dd, J = 8.8, 1.6 Hz, 1H), 7.72 (s, 1H), 7.48 (d, J = 7.2 Hz, 1H), 7.35−7.26 (m, 2H), 7.14 (d, J = 7.2 Hz, 2H), 7.02 (t, J = 7.6 Hz, 2H), 6.90 (t, J = 7.6 Hz, 1H), 6.84 (d, J = 8.0 Hz, 1H), 4.76 (dd, J = 9.6, 6.0 Hz, 1H), 4.22 (m, 1H), 4.11 (m, 2H), 4.0 (t, J = 10.0 Hz, 1H), 3.31−3.15 (m, 4H), 3.05 (dd, J = 13.2, 7.6 Hz, 1H), 2.96 (dd, J = 13.6, 7.2 Hz, 1H), 2.83 (dd, J = 14.0, 10.8 Hz, 1H), 2.19 (s, 3H), 1.92 (m, 1H), 0.92 (d, J = 6.8 Hz, 3H), 0.90 (d, J = 6.8 Hz, 3H); ^{13}C NMR (100 MHz, CDCl$_3$) δ 168.97, 158.34, 155.83, 153.43, 139.14, 137.98, 137.81, 135.84, 134.63, 129.87, 129.61 (2C), 128.58 (2C), 126.75, 125.08, 124.66, 122.56, 116.25, 113.81, 110.08, 72.63, 70.02, 58.96, 53.89, 53.80, 48.48, 36.02, 27.44, 24.89, 20.36, 20.11; HRMS (ESI) m/z calcd for C$_{33}$H$_{38}$N$_5$O$_7$S$_2$ [M + H]$^+$ 680.2213; found 680.2225.

(5S)-N-[(1S,2R)-3-[(6-Benzothiazolylsulfonyl)(2-methylpropyl)-amino]-2-hydroxy-1-(phenylmethyl)propyl]amino]carbonyl]-2-oxo-3-oxazolidinyl]phenyl]carbamic acid methyl ester (31d). To an ice-cooled solution of compound **29d** (0.1 g, 0.157 mmol) in CH$_2$Cl$_2$ (5 mL) was added an aqueous solution of Na$_2$CO$_3$ (0.026 g, 0.25 mmol) in water (2 mL) followed by the slow addition of methylchloroformate (12.3 μL, 0.157 mmol). After 15 min, the reaction mixture was warmed to ambient temperature and stirred until no starting material was detected by TLC (2 h). Reaction mixture was diluted with CH$_2$Cl$_2$, and layers were separated. The organic extract was washed with saturated aqueous NaCl solution (5 mL), dried (Na$_2$SO$_4$), filtered, and evaporated. The residue was purified by flash chromatography on silica gel, eluting with a mixture of 2% MeOH in EtOAc to afford the target compound **31d** (0.095 g, 87%) as a white solid. ^{1}H NMR (400 MHz, CDCl$_3$) δ 9.21 (s, 1H), 8.49 (d, J = 1.6 Hz, 1H), 8.25 (d, J = 8.8 Hz, 1H), 7.91 (dd, J = 8.8, 2.0 Hz, 1H), 7.76 (t, J = 1.6 Hz, 1H), 7.34−7.26 (m, 2H), 7.13 (d, J = 7.2 Hz, 2H), 7.06−6.89 (m, 6H), 4.76 (dd, J = 10.0, 6.0 Hz, 1H), 4.23 (m, 1H), 4.03 (t, J = 10.0 Hz, 2H), 3.82 (br s, 1H), 3.79 (s, 3H), 3.34 (dd, J = 9.2, 6.0 Hz, 1H), 3.25 (dd, J = 15.6, 9.2 Hz, 1H), 3.19−3.10 (m, 2H), 3.05 (dd, J = 13.6, 8.0 Hz, 1H), 2.94 (dd, J = 13.6, 7.2 Hz, 1H), 2.79 (dd, J = 13.6, 10.8 Hz, 1H), 1.90 (m, 1H), 0.94 (d, J = 6.8 Hz, 3H), 0.90 (d, J = 6.4 Hz, 3H); ^{13}C NMR (100 MHz, CDCl$_3$) δ 168.94, 158.31, 155.85, 154.23, 153.21, 139.12, 138.18, 137.57, 135.78, 134.63, 129.94, 129.57 (2C), 128.65 (2C), 126.82, 125.08, 124.69, 122.58, 114.88, 113.11, 108.87, 72.60, 69.94, 59.04, 53.89, 53.75, 52.73, 48.45, 35.88, 27.51, 20.36, 20.12. HRMS (ESI) m/z calcd for C$_{33}$H$_{38}$N$_5$O$_8$S$_2$ [M + H]$^+$ 696.2162; found 696.2158.

HIV-1 Protease Inhibition Assays. The HIV-1 protease inhibitory potencies of all compounds were determined by a fluorescence resonance energy transfer (FRET) method.[20,25,28] Protease substrate (Arg-Glu(EDANS)-Ser-Gln-Asn-Tyr-Pro-Ile-Val-Gln Lys-(DABCYL)-Arg) was purchased from Molecular Probes. The energy transfer donor (EDANS) and acceptor (DABCYL) dyes were labeled at two ends of the peptide, respectively, to perform FRET. Fluorescence measurements were carried out on a fluorescence spectrophotometer (Photon Technology International) at 30 °C. Excitation and emission wavelengths were set at 340 and 490 nm, respectively. Each reaction was recorded for about 10 min. The expression, isolation, and purification of wild-type and mutant HIV-1 protease used for binding experiments were carried out as previously described.[29] Wild-type HIV-1 protease (Q7K) and its MDR variant (L10I, L63P, A71V, G73S, I84V, L90M) were desalted through PD-10 columns (Amersham Biosciences). Sodium acetate (20 mM, pH 5) was used as the elution buffer. Apparent protease concentrations were around ~50 nM as estimated by UV spectrophotometry at 280 nm. All inhibitors were dissolved in dimethylsulfoxide (DMSO) and diluted to appropriate concentrations. Protease (2 μL) and inhibitor (2 μL) or DMSO were mixed and incubated for 20−30 min at room temperature before initializing the substrate cleavage reaction. For all experiments, 150 μL of a 1 μM substrate were used in substrate buffer [0.1 M sodium acetate, 1 M sodium chloride, 1 mM ethylenediaminetetraacetic acid (EDTA), 1 mM dithiothreitol (DTT), 2% DMSO, and 1 mg/mL bovine serum albumin (BSA) with an adjusted pH of 4.7]. Inhibitor binding dissociation constant (K_i) values were obtained by nonlinear regression fitting (GraFit 5, Erithacus software) of the plot of initial velocity as a function of inhibitor concentrations based on the Morrison equation.[30] The initial velocities were derived from the linear range of reaction curves.

Antiviral Assays. Drug susceptibility assays were carried out by Monogram Biosciences against three patient-derived strains of wild-type HIV-1 from clades A, B, and C and against two multidrug-resistant HIV-1 variants. Assays were carried out according to protocols detailed by Petropoulos et al.[31] and on the Web site http://www.monogrambio.com. Genbank accession numbers for the PR/RT regions of the six HIV-1 strains used in antiviral assays are as follows: WT-control, HQ179654; WT-A, HQ179655; WT-B, HQ179656; WT-C, HQ179657; MDR, HQ179658; MDR1, HQ179659.

Table 3. Crystallographic Data Collection and Refinement Statistics for Complexes of **17c** and **18d** with the Wild-Type HIV-1 Protease[a]

	17c	18d
resolution (Å)	1.95	1.85
space group	$P2_12_12_1$	$P2_12_12_1$
a (Å)	50.67	50.79
b (Å)	57.86	58.25
c (Å)	61.91	62.01
Z	4	4
R_{merge} (%)	10.2	8.7
completeness (%)	99.7	98.4
total no. of reflections	91 018	100 852
no. of unique reflections	13 809	16 073
R_{free} (%)	20.1	22.3
R_{factor} (%)	16.5	18.0
rmsd[a] in bond length (Å)	0.008	0.009
rms angle (deg)	1.340	1.317
temperature (°C)	−80	−80
PDB code	3MXD	3MXE

[a] rmsd, root-mean-square deviation; PDB, Protein Data Bank.

Protein Crystallography. The expression, isolation, and purification of wild-type and mutant HIV-1 proteases used for crystallization and binding experiments were carried out as previously described.[29] Cocrystals of the inhibitors with the wild-type protease were grown by hanging drop vapor diffusion method. The protease solution of concentration 1.2−1.8 mg/mL was equilibrated with 3-fold molar excess of inhibitors for at least 1 h prior to crystallization. The reservoir solution consisted of 126 mM phosphate buffer at pH 6.2, 63 mM sodium citrate, and ammonium sulfate in the range 24−29%.

The crystals used for data collection were mounted in Mitegen Micromounts and flash frozen over a nitrogen stream. Intensity data were collected at −80 °C on an in-house Rigaku X-ray generator equipped with an R-axis IV image plate. About 180 frames were collected per crystal with an angular separation of 1° and no overlap between frames. The data processing of the frames was carried out using the programs DENZO and ScalePack;[32,33] data collection statistics are listed in Table 3.

The crystal structures were solved and refined using the programs within the CCP4 interface.[34] Structure solution was obtained with the molecular replacement package AMoRe,[35] with 1F7A[36] as the starting model. When the solution was obtained, the molecular replacement phases were further improved using ARP/wARP[37] to build solvent molecules into the unaccounted regions of electron density. Model building was performed using the interactive graphics program Coot.[38] Conjugate gradient refinement using Refmac5[39] was performed by incorporating Schomaker and Trueblood tensor formulation of TLS (translation, libration, screw−rotation) parameters.[40−42] The working R (R_{factor}) and its cross validation (R_{free}) were monitored throughout the refinement. The refinement statistics are also shown in Table 3.

Supporting Information Available: Characterization data for additional target compounds and HPLC purity data for all compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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